

Chemistry 110 Study Guide

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and what you do not know as well as tell you where and what is in each chapter that is covered

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Definitions: Solution: a

homogeneous mixture composed of two or more substances Enthalpy:

the measure of energy in a

thermodynamic system State

Function: does not depend on the

path needed, but the beginning and

end Ion-ion: Ion-dipole: Dipole-

dipole: Ion induced dipole: Dipole

induced dipole: Hydrogen Bonding:

Ideal Gas: Vapor Pressure: Boiling

Point: Normal Boiling Point ... Chem

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STUDY GUIDE EXAM 3 November 7,

2016 FORM A ----- 1. A 2.60 g piece

of pure gold is exposed to 0.55 kJ of heat. If the gold is initially at 25.0 °C, is this enough energy to melt the sample? What is the resulting temperature of the gold? CHEMISTRY_110_STUDY_GUIDE_EXAM_3_for_d is - CHEMISTRY 110 ... Name:

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